FBD-12-2-23

Intermediate Part Second - 301

Objective Paper Code **8436**

CHEMISTRY (Objective) GROUP - II

Time: 20 Minutes

Marks: 17

Roll No.:

You have four choices for each objective type question as A, B, C and D. The choice which you think is correct, fill the relevant circle in front of that question number on computerized answer sheet. Use marker or pen to fill the circles. Cutting or filling two or more circles will result in zero marks in that question. Attempt as many questions as given in objective type question paper and leave other circles blank.coa

S.#	Questions	A	В	Ç	D
1	Formula of chloroform is:	CH₃Cℓ	CCℓ ₄	CH ₂ Cℓ ₂	CHCl3
2	Linear shape is associated with which set of hybrid orbitals?	sp	sp ²	sp ³	dsp ²
3	Group VI-B of transition elements contains;	Zn, Cd, Hg	Fe, Ru, Os	Ćr. Mo, W	Mn. Te, Re
4	Which one of these hydrogen halides is the weakest acid in solution?	HF	HBr	HI	НСℓ
5	One of the catalyst used in contact process.	Fe ₂ O ₃	V ₂ O ₅	SO ₃	Ag ₂ O
6	The chief ore of aluminum is:	Na ₃ AℓF ₆	Al ₂ O ₃ 2H ₂ O	$A\ell_2O_3$	$A\ell_2O_3 \cdot H_2O$
7	The oxide of beryllium is:	Acidic	Bàsic	Amphoteric	Composite
8	Keeping in view the size of atoms, which order is the correct one?	Mg≥Sr	Ва Мд	Lu > Ce	C£ > I
9	The normal amount of overhead ozone is about:	350 DU,	250 DU	150 DU	50 DU
10	Percentage composition of silica in cement is:	62	22	7:5	2\5
11	Succinic thiokinase is an example of:	Ligases	Lyases	Hydrolases	Isomerases
12	Flavour of amyl butyrate (Ester) is:	Orange	Apricot	Jasmine	Banana
13	The colour of precipitate of aldehyde with Fehling's solution is:	Black	White	Blue	Brick red
14	Rectified spirit contains ethyl alcohol about:	80%	85%.	90%	95%
15	Ethanol can be converted into ethanoic acid by:	Hydrogenation	Hydration	Oxidation	Fermentation
16	Which is not a nucleophile?	H ₂ O	H_2S	BF ₃	NH ₃
17	The electrophile in aromatic sulphonation is:	H ₂ SO ₄	rÍSO ₄	SO ₃	SO3 ⁺

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1.

Intermediate Part Secon' Roll No. CHEMISTRY GROUP - II (Subjective) Time: 02:40 Hours Marks: 68 SECTION - I 16 2. Write short answers to any EIGHT parts. Why CO2 is gas but SiO2 is a solid? Write four uses of borax. (ii) (iii) What is meant by chemical garden? (iv) Prepare glyoxal from benzene. (v) Give the structural formulae of (a) Benzoic acid (b) Acetophenone (vi) Write objections to Kekule formula of benzene. (vii) What are thermoplastic polymers? Give two examples. (viii) Give two points of difference between RNA and DNA. (ix) How is PVC prepared? Give its uses. What are primary and secondary pollutants? (x) (xi) How is water pollution caused by the detergents? (xii) What is dissolved oxygen? 3. Write short answers to any EIGHT parts. 16 Explain reforming of petroleum with the help of a suitable example. Define functional group. Give two examples of oxygen containing functional groups. (iii) Write structural formulae of these compounds: (a) 2,5-heptadiene (b) 1,3-pentadiene (iv) How can you chemically distinguish between propene and propyne? (v) Write the structural formula of the product formed when 1-butene reacts with Br_2 in $CC\ell_4$. (vi) Give reactions of HNO3 with reducing agents (a) FeSO4 (b) H2S (vii) How does concentrated H₂SO₄ react with (a) Copper (b) Ag. Give reactions. (viii) How does NO react with H2S and H2SO3? (ix) Write reaction of ethyl magnesium chloride with cyanogen chloride. (x) Give preparation of Grignard reagent in the presence of dry ether. (xi) How is potassium nitrate prepared on industrial scale? (xii) What products are formed in the pre-heating zone and decomposition zone of rotary kiln in cement industry? 12 4. Write short answers to any SIX parts. KMnO4 acts as oxidizing agents. Give reaction. Why does damaged tin plated iron get rusted quickly? (ii) Under what conditions does aluminium corrode? H₂C-CH₂ (iv) Give IUPAC names: (i) CH3-CH-COOH HÒ ÓH Why and how alcohol is denatured? (vi) How are ethers prepared by Williamsons synthesis?

(vii) How will you distinguish between methanal and ethanal by chemical reaction?

(viii) Give reaction of HCHO with NaBH4

(ix) What is ninhydrin test?

SECTION - II Attempt any THREE questions. Each question carries 08 marks.

5. (a) What are oxides? Discuss their classification on the basis of acidic and basic character.

(b) What is the role of gypsum in industry?

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6. (a) Write any eight applications of noble gases.(b) What is neutral sulphite semi-chemical process? Explain its bleaching unit.

7. (a) What is cracking? Discuss its types.

(b) Discuss Friedel-Crafts alkylation reaction of benzene with mechanism.

8. (a) Write any four methods for preparation of alkanes.

(b) Write any four methods for preparation of alkyl halides from alcohols.

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9. (a) Starting from aldehyde prepare: (i) Oxime (ii) Hydrazone (iii) Iodoform (iv) Cyanohydrin

(b) What types of reactions are shown by carboxylic acids? Describe any three reactions involving hydrogen atom of carboxylic acid group.