

FBD-12-18

Roll No. : _____

Objective
Paper Code

Intermediate Part Second (New Scheme)

CHEMISTRY (Objective)

8481

Time: 20 Minutes

Marks: 17



Q.No.1

You have four choices for each objective type question as A, B, C and D. The choice which you think is correct, fill the relevant circle in front of that question number on computerized answer sheet. Use marker or pen to fill the circles. Cutting or filling two or more circles will result in zero marks in that question. Attempt as many questions as given in objective type question paper and leave other circles blank.

| S.# | Questions | A | B | C | D |
|-----|--|--|--|-----------------------------------|---|
| 1 | Which statement is incorrect? | All the metals are good conductor of electricity | All the metals are good conductor of heat | All the metals form positive ions | All the metals form acidic oxides |
| 2 | Which does not belong to alkaline-earth metals? | Be | Ra | Ba | Rn |
| 3 | Chemical composition of colemanite is: | A | $\text{Ca}_2\text{B}_6\text{O}_{11} \cdot 5\text{H}_2\text{O}$ | C | $\text{Na}_2\text{B}_4\text{O}_7 \cdot 4\text{H}_2\text{O}$ |
| | | B | $\text{CaB}_4\text{O}_7 \cdot 4\text{H}_2\text{O}$ | D | $\text{CaNaB}_5\text{O}_9 \cdot 8\text{H}_2\text{O}$ |
| 4 | Laughing gas is chemically: | NO | N_2O | NO_2 | N_2O_4 |
| 5 | Which is the strongest acid: | HClO | HClO_2 | HClO_3 | HClO_4 |
| 6 | Which is a typical transition metal? | Sc | Y | Ra | Co |
| 7 | The state of hybridization of carbon in methane is: | sp^3 | sp^2 | sp | dsp^2 |
| 8 | Synthetic rubber is made by polymerization of: | Chloroform | Acetylene | Divinyl acetylene | Chloroprene |
| 9 | The electrophile in aromatic sulphonation is: | H_2SO_4 | HSO_4^- | SO_3 | SO_3^+ |
| 10 | Which is not a nucleophile? | H_2O | H_2S | BF_3 | NH_3 |
| 11 | Rectified spirit contains alcohol about: | 80% | 85% | 90% | 95% |
| 12 | Which compounds will not give iodoform test? | Acetaldehyde | Acetone | Butanone | 3-Pentanone |
| 13 | Which reagent is used to reduce a carboxylic acid to an alcohol? | LiAlH_4 | HI/P | H_2/Ni | H_2/Pt |
| 14 | The reaction between a fat and NaOH is called: | Esterification | Hydrogenolysis | Fermentation | Saponification |
| 15 | Which three elements are needed for the healthy growth of plants? | N, S, P | Na, Ca, P | N, P, K | N, K, C |
| 16 | Ecosystem is a smaller unit of: | Lithosphere | Biosphere | Atmosphere | Hydrosphere |
| 17 | A single chloride free radical can destroy how many ozone molecules? | 100 | 100000 | 10000 | 1000 |

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SECTION – I

2. **Write short answers to any EIGHT parts.** 16
- Why Na_2O is basic and P_2O_5 is acidic in character although both Na and P belong to same period?
 - Why second electron affinity value of an electron has positive sign?
 - What is milk of magnesia? Give its use.
 - Give any four uses of sodium silicate.
 - How “Al” (Aluminum) reacts with hydrogen and halogen?
 - Give four uses of Boric acid.
 - Why is SO_3 dissolved in H_2SO_4 and not in H_2O in contact process?
 - What is aqua regia? How it dissolves gold?
 - Give four dissimilarities between sulphur and oxygen.
 - How detergents are threat to aquatic animal life?
 - What is COD? How it is measured?
 - What is a functional group? Name functional group present in alcohol and ether.
3. **Write short answers to any EIGHT parts.** 16
- What are typical and non-typical transition elements?
 - What is coordination sphere? Give one example.
 - What is Clemmensen reduction? Give one example.
 - Convert propyne into acetone.
 - Benzene is polymer of acetylene. Justify.
 - Explain Wurtz synthesis with one example.
 - Explain Williamsons synthesis of ether.
 - Ethanol has higher boiling point than diethyl ether. Give reason.
 - Explain Fehling’s solution test.
 - Write two uses of formaldehyde.
 - What is Zwitter ion? Give example.
 - Write mechanism for the reaction between acetic acid and NH_3 ?
4. **Write short answers to any SIX parts.** 12
- Define saponification number with a suitable example.
 - Write two points of difference between a fat and an oil.
 - Differentiate with at least two points between amylose and amylopectin.
 - Name woody and non-woody raw materials for production of pulp (two each).
 - Describe the term setting of a cement.
 - Write formula for (a) Calcium super phosphate (b) Diammonium phosphate.
 - What is iodized salt?
 - Why iodine has metallic luster? Justify.
 - Name any two methods to manufacture bleaching powder. Also give reaction for this.

SECTION – II Attempt any THREE questions. Each question carries 08 marks.

5. (a) Write a brief note on oxidation state of elements in groups of modern periodic table? 04
 (b) Describe the role of lime in industry. Write eight points. 04
6. (a) How is KMnO_4 prepared by (i) Stadelers process (ii) Electrolytic oxidation process. 04
 (b) How is oil spillage affecting the marine life? 04
7. (a) Discuss cis-trans isomerism giving two examples. 04
 (b) Describe the stability of benzene on the basis of heat of hydrogenation. 04
8. (a) How is ethyne prepared by Kolbe’s electrolytic method? Write its mechanism. 04
 (b) How is ethanol prepared from molasses and starch by fermentation? 04
9. (a) Write the reactions of Grignard reagent with: (i) Alcohol (ii) CO_2 (iii) $\text{C}_6\text{H}_5\text{CN}$ (iv) CH_3COH 04
 (b) Write the reactions of acetaldehyde with: 04
- (i) $\text{NaBH}_4/\text{H}_2\text{O}$ (ii) H_2/Pd (iii) Dry $\text{HCl}/\text{C}_2\text{H}_5\text{OH}$ (iv) I_2/NaOH